

THERMAL PROPERTIES OF MATTER

Heat \Rightarrow It is form energy that when given to a body produces randomness in the body.

Heat possessed by a body is the total thermal energy of the body which is sum of kinetic energy of all the individual molecules of the body due to translational, vibrational and rotational motion of the molecule.

* SI unit of Heat is Joule and C.G.S unit is Erg.

$$1 \text{ Erg} = 10^{-7} \text{ Joule}$$

- * Heat is a scalar quantity.
- * Heat energy is part of internal energy in transit.
- * Heat energy is always transferred from high temperature to low temperature.

TEMPERATURE \Rightarrow Temperature is one of the seven fundamental quantity which decides the direction of flow of heat.

THERMOMETRY

The branch of physics that deals with the measurement of temperature is known as thermometry.

THERMOMETER \Rightarrow It is a device which is used to measure the temperature of body.

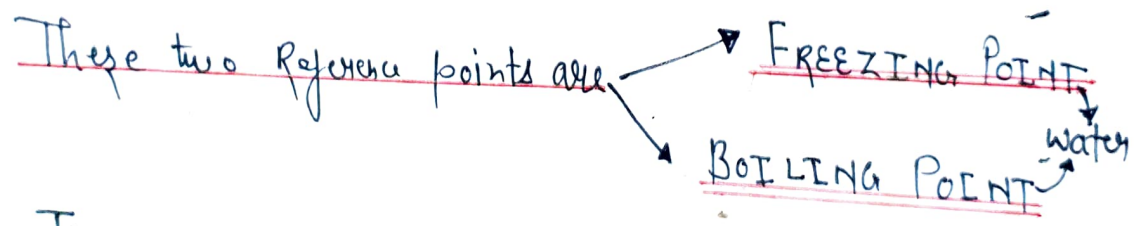
While designing thermometer we may take any of the following property that changes with change in temperature.

- (i) Variation of volume of liquid with temperature
- (ii) Variation of pressure or volume of gas with temperature
- (iii) Variation of resistance of metal with temperature.
- (iv) The variation of thermo emf with temperature of a junction in couple.

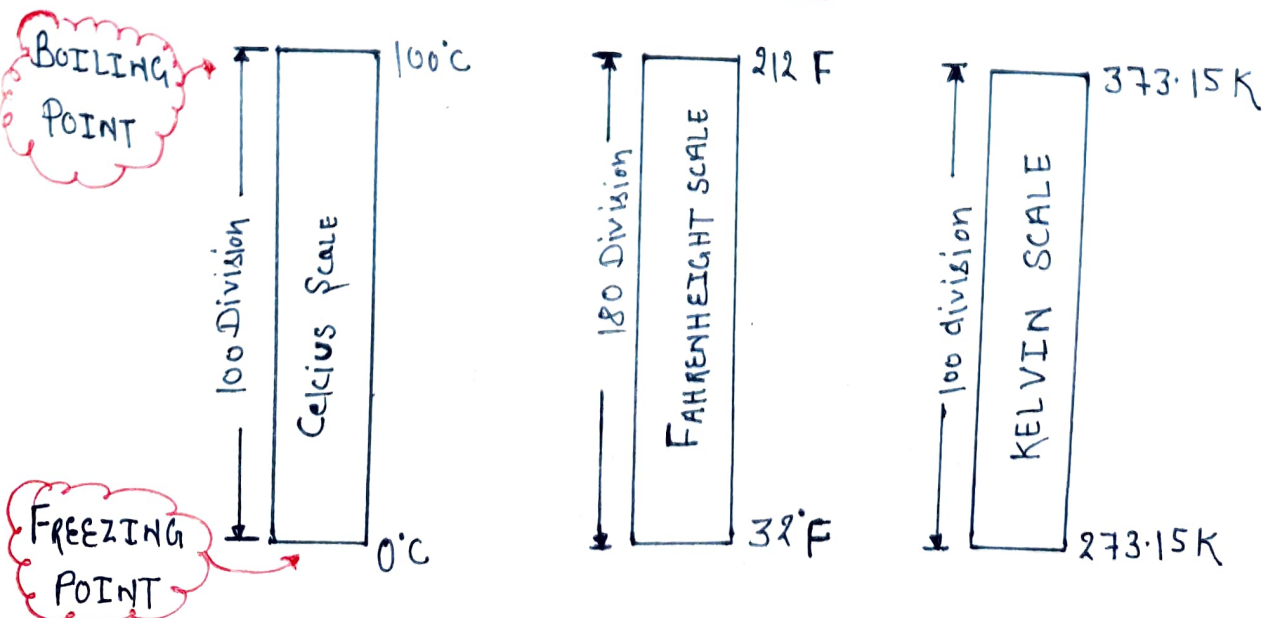
TEMPERATURE SCALE

Thermometer calibrated for a temperature scale is used to measure the value of given temperature on that scale.

In order to define any standard temperature scale, for the measurement of temperature we need two fixed reference point at which physical phenomena always occurs at same temperature.



THERE ARE FOLLOWING SCALE



One degree Celsius Change = One Kelvin Change = 1.8 Degree Fahrenheit Change

RELATIONSHIP BETWEEN DIFFERENT TEMPERATURE SCALE

1) Between °C and °F

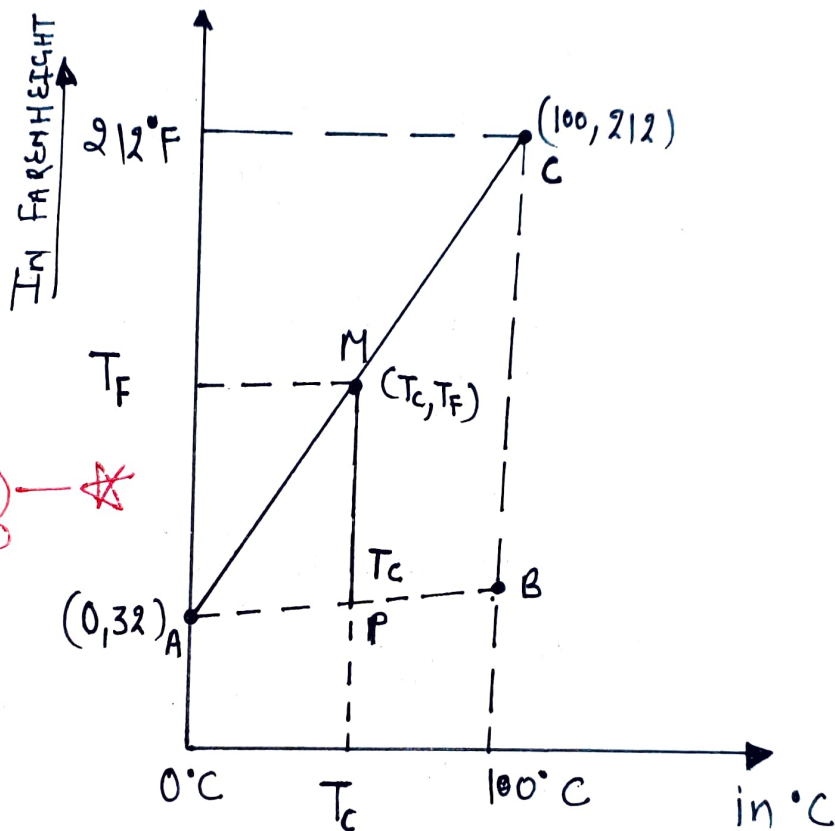
Since $\triangle ABC \sim \triangle APM$

So

$$\frac{AP}{AB} = \frac{MP}{BC}$$

$$\frac{T_C - 0}{100 - 0} = \frac{T_F - 32}{212 - 32}$$

$$\frac{T_C}{100} = \frac{T_F - 32}{180}$$



$$\frac{T_C}{5} = \frac{T_F - 32}{9}$$

→ Relationship Fahrenheit and degree Celsius

From * equation we can relate Any two scale.

$$\frac{\text{Temperature on scale (A)} - \text{FREEZING POINT}}{\text{Total No. Division [Difference b/w freezing and boiling point]}} = \frac{\text{Temperature on scale B} - \text{Freezing point}}{\text{Difference b/w freezing and boiling point}}$$

$$\frac{T_C - 0}{100 - 0} = \frac{T_F - 32}{212 - 32} = \frac{T_K - 273.15}{373.15 - 273.15}$$

$$\frac{T_C}{100} = \frac{T_F - 32}{180} = \frac{T_K - 273.15}{100}$$

$$\frac{T_C}{100} = \frac{T_K - 273.15}{100}$$

$$T_C = T_K - 273.15$$

Relationship b/w
°C and kelvin

NUMERICAL ON TEMPERATURE SCALE

Q → 1

Convert 30°C into kelvin and Fahrenheit

Sol

$$K = 273.15 + 30$$

$$K = 273.15 + 30 = 303.15 \text{ K}$$

$$K = 303.15 \text{ K}$$

$$\frac{T_C}{5} = \frac{T_F - 32}{9}$$

$$\frac{9}{5} T_C + 32 = T_F$$

$$\frac{9}{5} \times 30 + 32 = T_F$$

$$T_F = 86 \text{ F}$$

Q → 2 For Reumer's scale ice point is $0^{\circ}R$ and steam point is $80^{\circ}R$. Find the value of $20^{\circ}C$ in R .

Sol

$$\frac{T_R - \text{freezing point}}{\text{Steam point} - \text{Freezing point}} = \frac{T_C - \text{freezing point}}{\text{Steam point} - \text{Freezing point}}$$

$$\frac{T_R - 0}{80 - 0} = \frac{20 - 0}{100 - 0} \Rightarrow T_R = \frac{20 \times 80}{100}$$

$T_R = 16^{\circ}R$

Q → 3 If temperature is from $30^{\circ}C$ into $60^{\circ}C$ is changed then what is the change on kelvin scale and Fahrenheit scale?

Sol

We know:

$$\Delta 1^{\circ}C = \Delta 1^{\circ}K$$

∴ if $30^{\circ}C$ change then kelvin will also change by 30

For 30° change change in F $= 30 \times 1.8$ 54

OR

$$100^{\circ}C \text{ change} = 180^{\circ}F \text{ change}$$

$$1^{\circ}C \quad \quad \quad = \frac{180}{100}^{\circ}F$$

$$30^{\circ} \quad \quad \quad = \frac{180}{100} \times 30^{\circ}F \text{ change}$$

$$= 54^{\circ}F$$

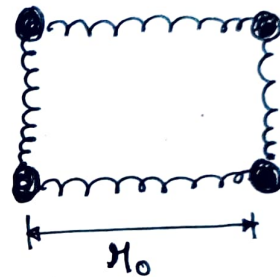
THERMAL EXPANSION

Whenever we supply heat to a substance then expansion in shape of substance will take place.

Reason Behind Expansion Of Substance.

→ Let us consider a substance in which molecules are at distance (r_0) when the energy was minimum (P.E₁). and molecules are at highly stable condition

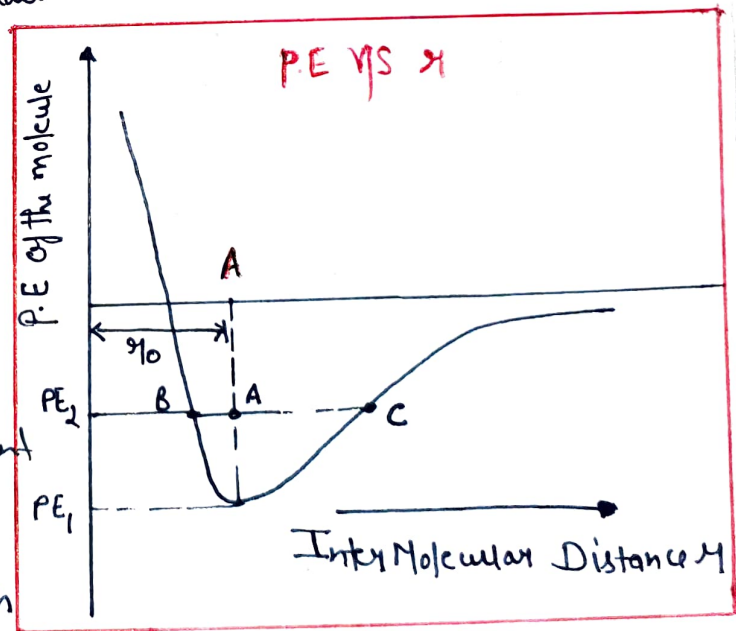
→ When we supply energy to the substance then energy of the molecule will increase to P.E₂. At this energy the distance b/w two molecules may have two values.



→ When we supply energy then molecules may come closer to each other then they will move away from each other.

→ The particles of molecule will start oscillating from mean position but with different displacement on both side.

→ We can see from graph that compression distance (AB) is less than expansion distance [AC]. So on an average we can see that substance will expand.



TYPES OF THERMAL EXPANSION

- ~~1. Linear Expansion~~
- ~~2. Area Expansion~~
- ~~3. Volume Expansion~~

LINEAR EXPANSION

When a rod of length L is heated such that its change in temperature ΔT then its length changes by ΔL .

Experimentally it was found that

$$\Delta L \propto L \quad \text{--- (1)}$$

$$\Delta L \propto \Delta T \quad \text{--- (2)}$$

Combining (1) and (2) equation

$$\Delta L \propto L \Delta T$$

$$\Delta L = \alpha L \Delta T$$

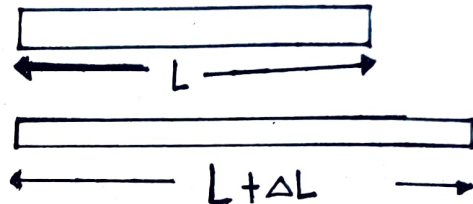
α = coefficient of Linear Expansion

$$\alpha = \frac{\Delta L}{L \Delta T}$$

if we take $L = 1$ $\Delta T = 1$ unit then

$$\alpha = \frac{\Delta L}{1}$$

So ⁶⁶ coefficient of thermal expansion is defined as small change in length per unit original length per unit change in temperature.



AREA EXPANSION

Let us consider a solid having surface Area A at temperature T . When heat is supplied to the solid such that its temperature changes by ΔT and its Area changes by ΔA .

The change in area was experimentally found to vary as follows

$$\Delta A \propto A \quad \text{--- (1)}$$

$$\Delta A \propto \Delta T \quad \text{--- (2)}$$

Combining equation (1) and (2)

$$\Delta A \propto A \Delta T$$

$$\Delta A = \beta A \Delta T$$

β = coefficient of area expansion

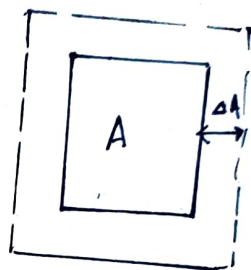
$$\beta = \frac{\Delta A}{A \Delta T}$$

if we take $A=1$, $\Delta T=1$

Then

$$\beta = \Delta A$$

Coefficient of ~~area~~ expansion is defined as ⁶⁶ change in surface area per unit original area per unit change in volume.⁹⁹



VOLUME EXPANSION [3-D Expansion]

Let us consider a solid having volume 'V' at temperature T. When the heat is supplied to the solid then temperature of the solid changed by ΔT and volume of solid changed by ' ΔV '. Experimentally it was found that volume expansion varies as follows

$$\Delta V \propto V \quad \text{--- (1)}$$

$$\Delta V \propto \Delta T \quad \text{--- (2)}$$

Combining (1) and (2)

$$\Delta V \propto V \Delta T$$

$$\Delta V = \gamma V \Delta T$$

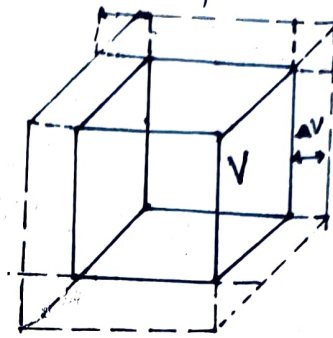
γ = Coefficient of volume Expansion

$$\gamma = \frac{\Delta V}{V \Delta T}$$

if $V=1$ and $\Delta T=1$

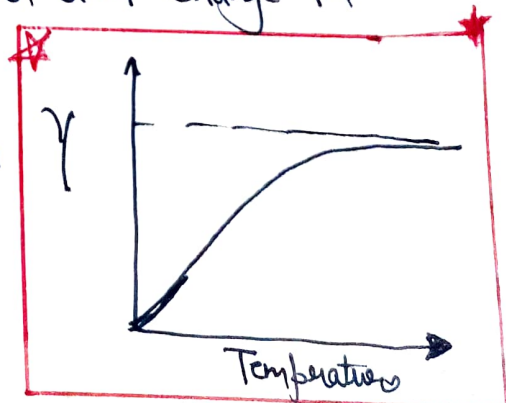
then

$$\gamma = \Delta V$$



Coefficient of volume expansion defined as "small change in volume per unit original volume per unit change in temperature."

- ★ At low temperature γ increases with temperature
- ★ At high temperature γ is independent to temperature and only depends upon Nature of material



RELATIONSHIP BETWEEN α , β and γ

Let us consider a cube having edge L , area of each face of cube A and volume of cube is V .

Now when heat is supplied to the cube and temperature of cube changes to ΔT then

$$\text{New length of cube } L' = L + \alpha L \Delta T$$

$$\text{New Area of face of cube } A' = A + \beta A \Delta T$$

$$\text{New volume of cube } = V' = V + \gamma V \Delta T$$

(I) Relationship b/w α and β

$$A' = A + \beta A \Delta T = (L + \Delta L)^2$$

$$A + \beta A \Delta T = (L + \alpha L \Delta T)^2$$

$$L^2 = A$$

$$A(1 + \beta \Delta T) = L^2 (1 + \alpha \Delta T)^2$$

$$A(1 + \beta \Delta T) = A [1 + 2\alpha \Delta T + \alpha^2 \Delta T^2]$$

$$1 + \beta \Delta T = 1 + 2\alpha \Delta T$$

$$\beta \Delta T = 2\alpha \Delta T$$

$\left\{ \begin{array}{l} \alpha^2 \Delta T^2 \text{ is} \\ \text{very small term} \\ \text{so it can be} \\ \text{neglected} \end{array} \right\}$

$$\beta = 2\alpha \Rightarrow \alpha = \frac{\beta}{2} \quad \text{--- (1)}$$

(II) Relationship b/w α and γ

$$V' = V + \gamma V \Delta T = (L + \Delta L)^3$$

$$V(1 + \gamma \Delta T) = (L + \alpha L \Delta T)^3$$

$$\left\{ \begin{array}{l} L^3 = V \end{array} \right.$$

$$V(1 + \gamma \Delta T) = L^3 (1 + \alpha \Delta T)^3$$

$$V(1 + \gamma \Delta T) = V [1 + 3\alpha \Delta T + 3\alpha^2 \Delta T^2 + \alpha^3 \Delta T^3]$$

$$1 + \gamma \Delta T = 1 + 3\alpha \Delta T$$

$$\gamma \Delta T = 3\alpha \Delta T$$

$$\gamma = 3\alpha$$

$$\alpha = \frac{\gamma}{3} \quad \text{--- } \alpha$$

from equation (1) and (2)

$$\alpha = \frac{\beta}{2} = \frac{\gamma}{3}$$

$3\alpha^2 \Delta T^2$
 $\alpha^3 \Delta T^3$ → These terms are very small & 0 they can be neglected

NUMERICAL

Q.1

For an ideal gas find the value of γ at constant pressure when temperature is changed.

Sol

$$PV = nRT$$

$$Pdv = nRdT$$

$$Pv\gamma dT = nRdT$$

$$\left\{ \Delta V = \gamma V dT \right.$$

$$\gamma = \frac{nR}{Pv}$$

$$\gamma = \frac{nR}{nRT}$$

$$\gamma = \frac{1}{1}$$

Numerical

EXPANSION OF BINETALLIC STRIP

Let us consider two strip of two different material rivited together

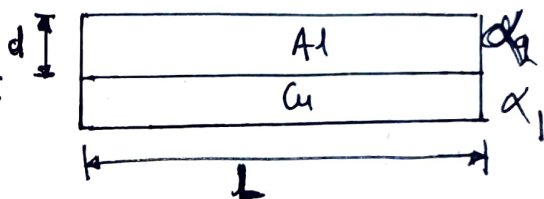
	Length	Thickness	Coefficient of linear expansion
Copper strip →	L	d	α_1
Aluminium strip →	L	d	α_2

If $\alpha_2 > \alpha_1$ and the strip combination is heated so that temperature change by ΔT so change in length of the two strips take place

$$L_1 = L(1 + \alpha_1 \Delta T)$$

$$L_2 = L(1 + \alpha_2 \Delta T)$$

Since $L_2 > L_1$ because $\alpha_2 > \alpha_1$ the strip will bend to form an arc of Radius R



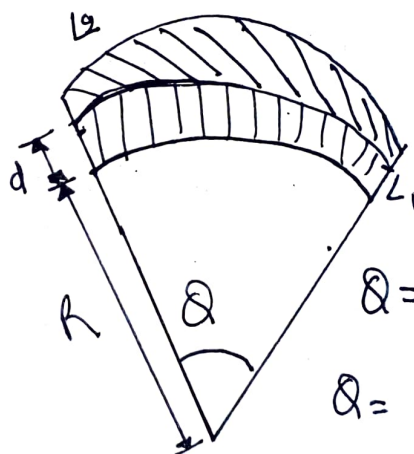
from equation (1) and (2)

$$\frac{L_2}{R+d} = \frac{L_1}{R}$$

$$RL_2 = L_1R + L_1d$$

$$R[L_2 - L_1] = L_1d$$

$$R = \frac{L_1d}{L_2 - L_1}$$



$$\theta = \frac{\text{Length of arc}}{\text{Radius}}$$

$$\theta = \frac{L_1}{R} \quad \text{--- (1)}$$

$$\theta = \frac{L_2}{R+d} \quad \text{--- (2)}$$

$$R = \frac{L(1 + \alpha_1 \Delta T)d}{L(1 + \alpha_2 \Delta T) - L(1 + \alpha_1 \Delta T)}$$

$$R = \frac{L(1 + \alpha_2 \Delta T)d}{L[1 + \alpha_2 \Delta T - 1 - \alpha_1 \Delta T]}$$

$$\Rightarrow R = \frac{(1 + \alpha_2 \Delta T)d}{[\alpha_2 - \alpha_1] \Delta T}$$

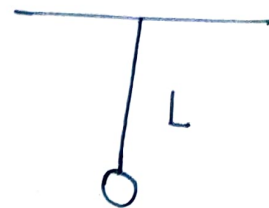
Numerical

FAULT IN TIME PERIOD OF PENDULUM CLOCK

Let us consider a pendulum clock in which bob is hanged by a metallic strip of length L having coefficient of linear expansion α at temperature θ

∴ Time period of pendulum at temperature θ

$$T = 2\pi \sqrt{\frac{L}{g}} \quad \text{--- (1)}$$

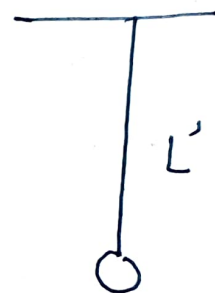


Now when the temperature of pendulum is changed then length of pendulum changes to L'

$$L' = L(1 + \alpha \Delta \theta)$$

∴ New time period

$$T' = 2\pi \sqrt{\frac{L'}{g}} \quad \text{--- (2)}$$



$\Delta \theta =$ change in temp

Divide (2) by (1)

$$\frac{T'}{T} = \left(\frac{L'}{L} \right)^{1/2}$$

$$T' = T \left[\frac{L(1 + \alpha \Delta \theta)}{L} \right]^{1/2}$$

$$T' = T (1 + \alpha \Delta \theta)^{1/2}$$

$$T' = T \left(1 + \frac{1}{2} \alpha \Delta \theta \right)$$

expanding binomially
 $(1+x)^n = 1 + nx$

When $T =$ Original Time period

$\Delta \theta =$ Change in temperature

$T' =$ New time period.

TRANSFER OF HEAT

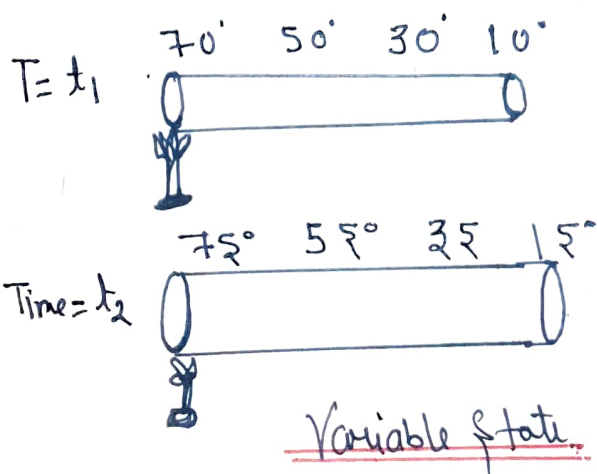
Heat is a form of energy which can be transferred from one part of the system to another part depending upon the difference of temperature b/w them.

Methods Of Heat TRANSFER

I CONDUCTION II Convection III Radiation.

VARIABLE STATE

- If the molecules of rod are absorbing some portion of heat supplied to it and remaining heat is transferred to other molecule then it is known as variable state
- In this different part of rod will have different temperatures with respect to time and distance
- Heat passing through each cross-section will not be same.

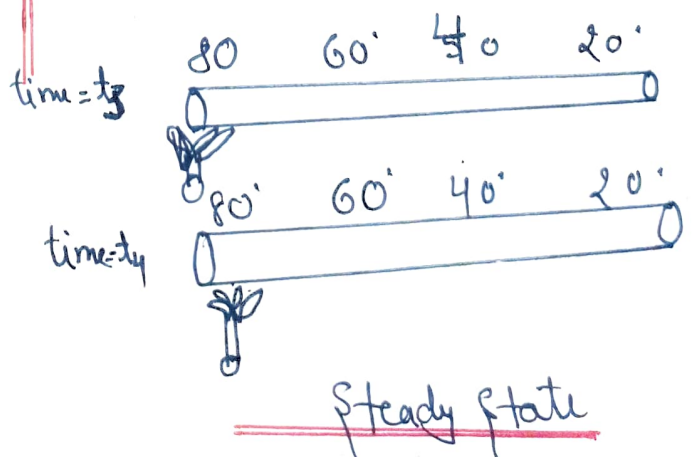


STEADY STATE

If the molecules of the rod are transferring all the heat to the other molecule and they do not absorb any amount of heat is known as steady state.

In this different part of rod will have constant temperature ~~and~~ and temperature will not change with time.

In this heat passing through each cross-section will be same.



... proportional to temperature difference

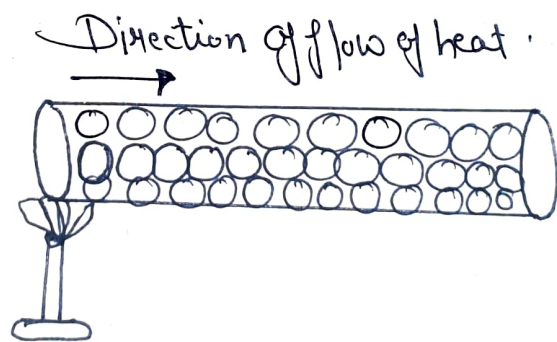
Equation Of Steady State

$$\left(\frac{dT}{dt}\right)_{x=\text{constant}} = 0$$

This means that temperature of rod at a given distance is constant. i.e. temperature does not change with time.

CONDUCTIVITY

It is a mode of heat transfer from one part of the body to another, from particle to particle in the direction of fall of temperature without any actual movement of the heated particle.



In this heat is transferred because of vibration of molecule.

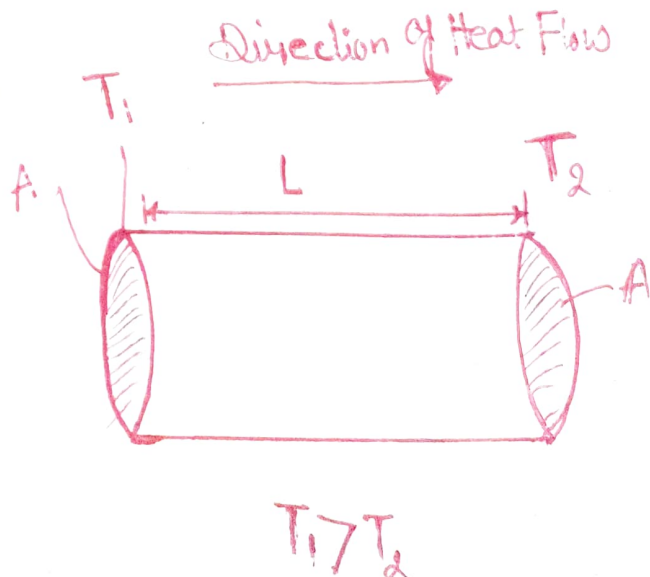
THERMAL CONDUCTIVITY

Let us consider a rod having length 'L' and cross-sectional area A and the temperature at its two ends is T_1 and T_2 respectively.

In steady state the heat flowing through rod in time t is found to

(1) It is directly proportional to area of cross-section

$$Q \propto A \quad \text{--- (1)}$$



~~(i)~~ It is directly proportional to temperature difference

$$Q \propto T_1 - T_2$$

$$Q \propto \Delta T \quad \text{--- (2)}$$

$$\{\Delta T = T_1 - T_2\}$$

~~(ii)~~ It is inversely proportional to the length b/w two ends of rod

$$Q \propto \frac{1}{L} \quad \text{--- (3)}$$

$$\text{(iv)} \quad Q \propto t \quad \text{--- (4)}$$

It is directly proportional to time.

Combining (1), (2), (3) and (4) equation

$$Q \propto \frac{A \Delta T}{L}$$

$$\Rightarrow Q = \frac{KA \Delta T t}{L}$$

$$Q = \frac{KA \Delta T t}{L}$$

Where K = Coefficient of thermal conductivity

SI unit of K = $\text{Joule s}^{-1} \text{m}^{-1} \text{K}^{-1}$

$\text{W m}^{-1} \text{K}^{-1}$

C.G.S unit = $\text{Cal} \cdot \text{s}^{-1} \text{cm}^{-1} \cdot \text{C}^{-1}$

Heat Current (H)

It is the rate of transfer of Heat.

$$H = \frac{Q}{t} = \frac{KA \Delta T t}{L t}$$

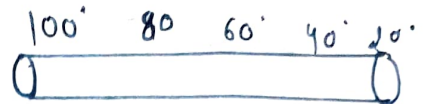
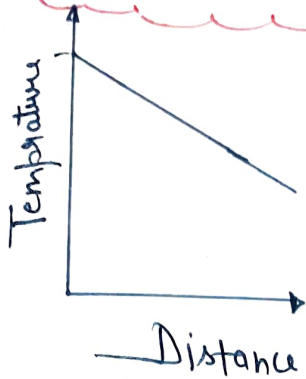
$$H = \frac{KA \Delta T}{L}$$

TEMPERATURE GRADIENT

It is defined as change in temperature with respect to change in distance.

$$\text{Temperature Gradient} = \frac{dT}{dx}$$

$$\frac{dT}{dx} = \text{slope}$$



So slope of the graph b/w temperature and distance give temperature gradient.

ANALOGY

Heat

Heat Current $\rightarrow H$

Heat flows due temperature difference ΔT

Flow of Heat is opposed by Thermal Resistance.

$$H = \frac{KA\Delta T}{l}$$

$$H = \frac{\Delta T}{\frac{l}{KA}}$$

$$H = \frac{\Delta T}{R_{th}}$$

R_{th} = Thermal Resistance

Electricity

Current $- I$

Current flows due to ~~temp~~ potential difference V

The flow of current is opposed by Resistance R

$$V = IR$$

$$I = \frac{V}{R}$$

$$\begin{array}{ccc}
 H & \longrightarrow & I \\
 \Delta T & \longrightarrow & V \\
 R_{Th} & \longrightarrow & R
 \end{array}$$

$$R_{Th} = \frac{l}{KA}$$

R_{Th} = Thermal Resistance.

Important Formula

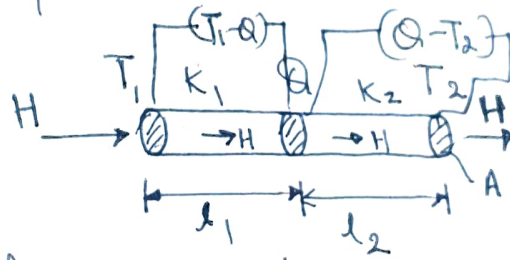
$$H = \frac{\Delta T}{R_{Th}}$$

$$R_{Th} = \frac{l}{KA}$$

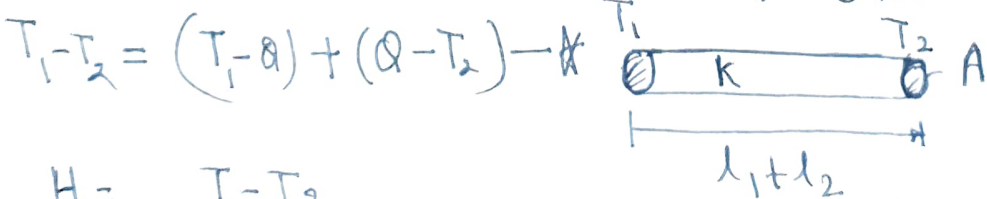
$$Q = \frac{KA \Delta T l}{L}$$

Equivalent Thermal Conductivity

In series combination same ~~current~~ heat will flow through conductor but ~~the~~ temperature difference will be different



equivalent conductor



$$H = \frac{T_1 - T_2}{R_{Th}}$$

$$T_1 - T_2 = H(R_{Th}) \quad \text{--- (1)}$$

$$T_1 - Q = H(R_{Th1}) \quad \text{--- (2)}$$

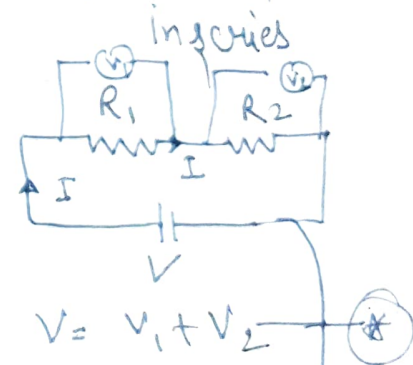
$$Q - T_2 = H(R_{Th2}) \quad \text{--- (3)}$$

Putting the value of (2) and (3) in (1)

$$H(R_{Th}) = H(R_{Th1}) + H(R_{Th2})$$

$$R_{Th} = R_{Th1} + R_{Th2}$$

Equivalent Resistance



$$V = V_1 + V_2$$

$$V = I R_{eq} \quad \text{--- (1)}$$

$$V_1 = I R_1 \quad \text{--- (2)}$$

$$V_2 = I R_2 \quad \text{--- (3)}$$

Put the value of (2) and (3) in equation (1)

$$I R_{eq} = I R_1 + I R_2$$

$$R_{eq} = R_1 + R_2$$

$$R_{TH} = \frac{l}{KA}$$

$$R_{TH1} =$$

$$\frac{l_1}{K_1 A}$$

$$R_{TH2} = \frac{l_2}{K_2 A}$$

$$\frac{L}{KA} = \frac{l_1}{K_1 A} + \frac{l_2}{K_2 A}$$

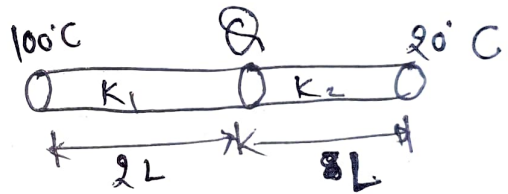
$$\Rightarrow \left(\frac{L}{K} = \frac{l_1}{K_1} + \frac{l_2}{K_2} \right)$$

When $L =$ Total length of conductor,
 $K =$ equivalent thermal conductivity

Q → Find equivalent thermal conductivity for the following.

Sol

$$\frac{L}{K} = \frac{L_1}{K_1} + \frac{L_2}{K_2}$$



$$\frac{2L+L}{K} = \frac{2L}{K_1} + \frac{L}{K_2}$$

$$\frac{3L}{K} = L \left(\frac{2}{K_1} + \frac{1}{K_2} \right)$$

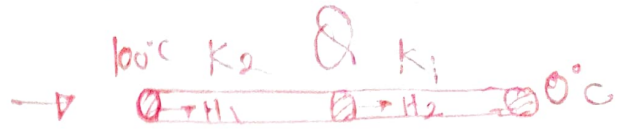
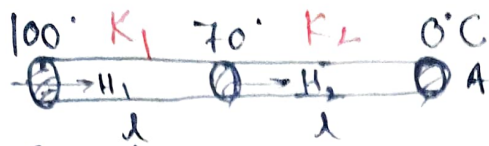
$$\frac{3}{K} = \left(\frac{2K_2 + K_1}{K_1 K_2} \right)$$

$$\frac{K}{3} = \frac{K_1 + K_2}{2K_2 + K_1}$$

$$\Rightarrow K = \frac{3(K_1 + K_2)}{2K_2 + K_1}$$

Q →

Find the temperature of the junction when position of conductor is reversed.



Sol

Before Reversing
 $H_1 = H_2$

$$\frac{\Delta T}{R_1} = \frac{\Delta T}{R_2}$$

$$\frac{100-70}{\frac{l}{K_1}} = \frac{70-0}{\frac{l}{K_2}}$$

$$\frac{30K_1}{l} = \frac{70K_2}{l}$$

$$\frac{K_1}{K_2} = \frac{70}{30} = \frac{7}{3}$$

$$\frac{K_1}{K_2} = \frac{7}{3}$$

After Reversing.

$$H_1 = H_2$$

$$\frac{\Delta T}{R_1} = \frac{\Delta T}{R_2}$$

$$\frac{100-Q}{\frac{l}{K_2}} = \frac{Q-0}{\frac{l}{K_1}}$$

$$K_2(100-Q) = K_1(Q-0)$$

$$100-Q = \frac{K_1}{K_2} Q$$

$$100-Q = \frac{7}{3} Q$$

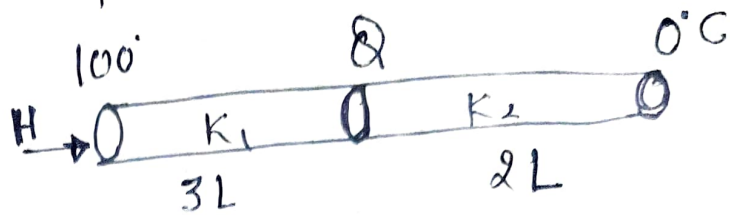
$$100 = \frac{7}{3} Q + Q$$

$$100 = \frac{10Q}{3}$$

$$Q = 30^\circ \text{C}$$

Q → Find the junction temperature if $k_1 = 2k_2$

Sol Let the junction temperature be Q



Since the heat through both conductors is same

$$H = \frac{\Delta T}{R_1} = \frac{\Delta T}{R_2}$$

$$\frac{100 - Q}{R_1} = \frac{Q - 0}{R_2}$$

$$R_2(100 - Q) = R_1 Q$$

$$\frac{2L}{k_2}(100 - Q) = \frac{3L}{k_1} Q$$

$$\frac{2L}{k_2}(100 - Q) = \frac{3L}{k_1} Q$$

$$\frac{2}{k_2}(100 - Q) = \frac{3}{2k_2} Q$$

$$100 - Q = \frac{3}{4} Q$$

$$100 = \frac{3Q + Q}{4}$$

$$100 = \frac{7Q}{4}$$

$$Q = \frac{400}{7} = 57.1$$

$$Q = 57.1^\circ\text{C}$$

Q →

Find the temperature of the junction when position of conductor is reversed.



Sol

Before Reversing
 $H_1 = H_2$

$$\frac{\Delta T}{R_1} = \frac{\Delta T}{R_2}$$

$$\frac{100-70}{\frac{\lambda}{K_1}} = \frac{70-0}{\frac{\lambda}{K_2}}$$

$$\frac{30K_1}{\lambda} = \frac{70K_2}{\lambda}$$

$$\frac{K_1}{K_2} = \frac{70}{30} = \frac{7}{3}$$

$$\frac{K_1}{K_2} = \frac{7}{3}$$

After Reversing.

$$H_1 = H_2$$

$$\frac{\Delta T}{R_1} = \frac{\Delta T}{R_2}$$

$$\frac{100-Q}{\frac{\lambda}{K_2}} = \frac{Q-0}{\frac{\lambda}{K_1}}$$

$$K_2(100-Q) = K_1(Q-0)$$

$$100-Q = \frac{K_1}{K_2} Q$$

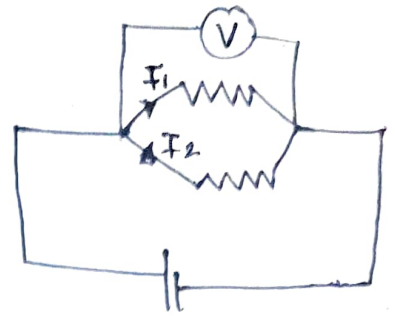
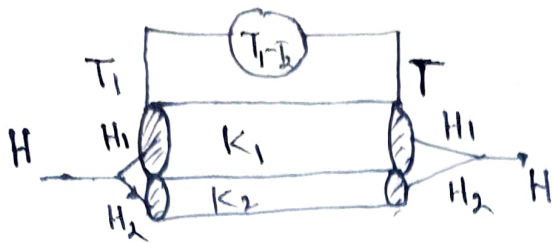
$$100-Q = \frac{7}{3} Q$$

$$100 = \frac{7}{3} Q + Q$$

$$100 = \frac{10}{3} Q$$

$$Q = 30^\circ \text{C}$$

Equivalent Thermal Conductivity In Parallel Combination



$$\frac{1}{R_p} = \frac{1}{R_1} + \frac{1}{R_2}$$

Using Analogy with electrical
Circuit
Equivalent ^{thermal conductivity} ~~conductivity~~ in parallel
Combination

$$\frac{1}{R_{PTH}} = \frac{1}{R_{TH(1)}} + \frac{1}{R_{TH(2)}} \quad \text{--- (*)}$$

$$R_{PTH} = \frac{l}{KA} \quad \text{--- (1)} \quad R_{TH(1)} = \frac{l}{K_1 A_1} \quad \text{--- (2)}$$

$$R_{TH(2)} = \frac{l}{K_2 A_2} \quad \text{--- (3)}$$

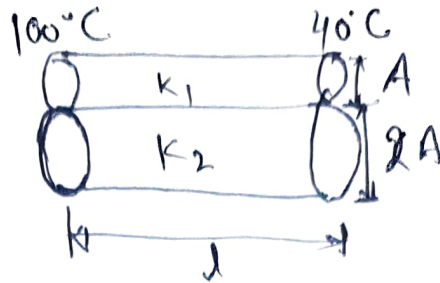
Substituting the value of (1), (2) and (3) in *

$$\frac{KA}{l} = \frac{K_1 A_1}{l} + \frac{K_2 A_2}{l}$$

$$K = \frac{K_1 A_1 + K_2 A_2}{A}$$

Question Based Upon Parallel Combination.

Q → Find Equivalent ^{Thermal} Resistance and equivalent Thermal Conductivity



Sol

$$R_1 = \frac{l}{k_1 A}$$

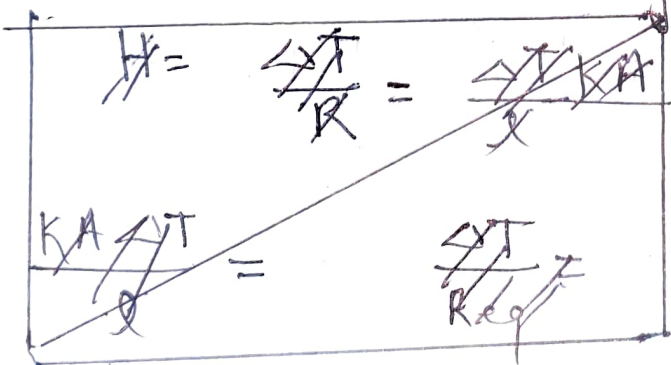
$$R_2 = \frac{l}{k_2 \cdot 2A}$$

$$\frac{1}{R} = \frac{1}{R_1} + \frac{1}{R_2} = \frac{1}{R} = \frac{2k_2 A}{l} + \frac{k_1 A}{l}$$

$$\frac{1}{R} = \frac{2k_2 A + k_1 A}{l}$$

$R_{eq} = \frac{l}{A(2k_2 + k_1)}$
→ Equivalent Thermal Resistance

Equivalent Thermal Conductivity



$$R_{eq} = \frac{l}{k_{eq} \text{ total area}}$$

$$\frac{l}{A(2k_2 + k_1)} = \frac{l}{k_{eq}}$$

$$A(2k_2 + k_1) = k_{eq}$$

$$k_{eq} = A(2k_2 + k_1)$$

$$R_{eq} = \frac{l}{A(2k_2 + k_1)}$$