

## ASSIGNMENT NO. 6

SUBJECT: PHYSICS CLASS-XI DEC,2025

## CH: 10 Thermal properties of Matter

- 1. What is meant by coefficient of linear expansion, superficial expansion and cubical expansion? derive the relation between them.
- 2. Prove that the coefficient of cubical expansion of an ideal gas at constant pressure is equal to the reciprocal of its absolute temperature.
- 3. Define coefficient of thermal conductivity. Write its S.I unit.
- **4.** A copper block of mass 2.5 kg is heated in a furnace to a temperature of 500 °C and then placed on a large ice block. What is the maximum amount of ice that can melt? (Specific heat of copper = 0.39 J g<sup>-1</sup> K<sup>-1</sup>; heat of fusion of water= 335 J g<sup>-1</sup>).
- 5. State Wein's displacement law.
- 6. In an experiment on the specific heat of a metal, a 0.20 kg block of the metal at 150 °C is dropped in a copper calorimeter (of water equivalent 0.025 kg) containing 150 cm<sup>3</sup> of water at 27 °C. The final temperature is 40 °C. Compute the specific heat of the metal.

## **CH: 11 Thermodynamics**

- 7. Zeroth law of thermodynamics.
- 8. First law of thermodynamics.
- 9. Second law of thermodynamics.
- 10. Derive and expression for work done in an isothermal process by an ideal gas.
- 11. Derive a formula for the work done by an ideal gas in an adiabatic process.
- 12. Derive a relation between two principle specific heats of a gas or derive Mayer's formula.
- 13. Show that slope in adiabatic process is  $\gamma$  times the slope in isothermal process.

## **CH: 12 Kinetic Theory**

- 14. Derive an expression for the pressure due to an ideal gas.
- 15. Kinetic interpretation of temperature.
- 16. State the law of equipartition of energy.
- 17. Defines degree of freedom. Calculate the degrees of freedom of monoatomic, diatomic and triatomic gas molecules.
- **18.** What is meant by mean free path of a gas molecule? Derive an expression for it. On which factors does it depend?